Evaluation of covalent inhibition potency: From IC_{50} to k_{inact}/K_I and beyond

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- I. Foundations: Theory and practice in covalent inhibition kinetics
	- Kinetic mechanisms of covalent enzyme inhibition
	- Measures of covalent inhibitory potency
	- \bullet A close look at the covalent IC_{50}
- II. Applications: Three enzymology examples
	- Inhibition of an E3 ligase: A clear failure of the IC_{50} method.
	- **Inhibition of the JAK3 kinase:** Power of the IC_{50} method is open to discussion.
	- Inhibition of an EGFR mutant: Beyond the k_{inact}/K_I .
- III. Software: DynaFit / DynaPlate automation package
	- Inhibition of Bruton tyrosine kinase: A live software demo.
- **IV. Open Discussion**

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Kinetic mechanisms of covalent enzyme inhibition

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Microscopic rate constants (from continuous real-time assays)

- Mechanism "C2S": k_a , k_d , k_f
- \bullet Mechanism "C2F": k_d , k_f [Note: k_a is *fixed* at a "diffusion controlled" value]
- \bullet Mechanism "C1": k_a

Derived kinetic constants

Mechanisms "C2S" and "C2F"

\n- $$
k_{\text{inact}} = k_f
$$
\n- $K_i = k_d / k_a$
\n- $K_I = (k_d + k_f) / k_a$
\n- $k_{\text{inact}} / K_I = k_a k_f / (k_d + k_f)$
\n

- Mechanism "C1"
	- $k_{\text{inact}}/K_I = k_a$
- \bullet The IC₅₀ (from end-point assays)

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- \bullet Multiple definitions of covalent IC_{50} in the literature
- \bullet Time-dependence of covalent IC₅₀
- \bullet Dependence of covalent IC₅₀ on other experimental factors

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Repeat at various inhibitor concentrations (including zero):

- Method $1:$ $[1]$
	- \bullet Incubate enzyme $+$ covalent inhibitor for a specific duration of time
	- Stop the covalent reaction somehow
	- Add substrate to determine amount of residual free enzyme
	- Compute the initial reaction rate since substrate was was added
- Method 2: [2]
	- Incubate enzyme + covalent inhibitor + substrate for a specific duration of time
	- Stop the enzymatic reaction
	- Determine the amount of product

Method $3:$ $^{[3]}$

- \bullet Incubate enzyme $+$ covalent inhibitor for a specific duration of time
- Add substrate
- Continue incubating enzyme $+$ inhibitor $+$ substrate for another fixed time interval
- Stop the enzymatic reaction and the covalent $E + I$ reaction
- Determine the amount of product

[1] Kitz & Wilson (1962)

[2] Krippendorff et al. (2009)

 $\overline{31}$ Fassunke *et al.* (2018)

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Covalent IC_{50} is by definition time-dependent

A simulation study to illustrate time-dependence of IC_{50} :

Figure 1. Dose response illustration of the time dependency of IC_{50} for covalent inhibitors (model data).

Thorarensen, A. et al. (2021) Bioorg. Med. Chem. 29, 115865.

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Fassunke, J. et al. (2018) Nature Commun. 9, 4655.

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 IC_{50} [nM]

 \leq 1

 <1

 1.3 ± 0.1

 <1

 2.1 ± 1.0

- As incubation time approaches infinity, all covalent inhibitors end up with the same $IC_{50}!$
- **•** Let us figure out together what it is.

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- **II. Applications: Three enzymology examples**
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An E3 ligase can be conveniently assayed in the absence of E1 and E2.

Fluorescence polarization decreases as the fluorophore is detached from ubiquitin.

Krist, D. T. et al. (2016) Chem. Sci. 7, 5587–5595

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E3 ligase assay: Typical covalent inhibition datasets

• The observed decrease of fluorescence polarization was inverted for convenience.

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[mechanism]
E + S ---> E + P : kaS
E + I \leq = \geq E.I : ka kd
E I ---> E-I \cdot kf[constants]
k = 0.0059897ka = 100kd = \{0.01, 0.1, 1, 10, 100\} ??
kf = \{0.000001, 0.00001, 0.0001, 0.001, 0.01, 0.1\} ??
```
- **•** First-order substrate kinetics $(|S| << K_m)$.
- Two-step inhibitor binding ("rapid equilibrium" in the initial binding step).

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$$
\frac{\mathrm{d}[E]}{\mathrm{d}t} = -k_{\mathrm{a}}[E][I] + k_{\mathrm{d}}[E.I] \tag{1}
$$

$$
\frac{\mathrm{d}[\mathrm{S}]}{\mathrm{d}t} = -k_{\mathrm{aS}} \left[\mathrm{E} \right] [\mathrm{S}] \tag{2}
$$

$$
\frac{\mathrm{d}[P]}{\mathrm{d}t} = +k_{\mathrm{aS}}\left[\mathrm{E}\right][S] \tag{3}
$$

$$
\frac{\mathrm{d}[\mathrm{I}]}{\mathrm{d}t} = -k_{\mathrm{a}}\left[\mathrm{E}\right][\mathrm{I}] + k_{\mathrm{d}}\left[\mathrm{E}.\mathrm{I}\right] \tag{4}
$$

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\frac{\mathrm{d}[E.I]}{\mathrm{d}t} = +k_{a}[E][I] - k_{d}[E.I] - k_{f}[E.I]
$$
\n(5)

$$
\frac{\mathrm{d}[E-I]}{\mathrm{d}t} = +k_{\mathrm{f}} [E.I] \tag{6}
$$

This mathematical model was auto-generated by DynaFit software from [mechanism].

- **Pirack** best-fit model parameters are the **optimized** microscopic rate constants:
	- k_d dissociation rate constant
	- $\frac{k_{\rm f}}{k_{\rm a}}$ forward isomerization rate constant
		- association rate constant is a fixed model parameter
- Derived best-fit model parameters are the macroscopic kinetic constants:

$$
K_{\rm i} = \frac{k_{\rm d}}{k_{\rm a}} \tag{7}
$$

$$
K_{\rm I} = \frac{k_{\rm d} + k_{\rm f}}{k_{\rm a}} \tag{8}
$$

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k_{\rm inact} = k_{\rm f} \tag{9}
$$

$$
k_{\rm inact}/K_{\rm I} = k_{\rm f} \frac{k_{\rm a}}{k_{\rm d} + k_{\rm f}} \equiv k_{\rm eff} \tag{10}
$$

 \bullet " k_{eff} " a.k.a. k_{inact}/K_I is the covalent efficiency constant.

E3 ligase assay: Covalent-kinetic correlation (CKC)

• The k_{inact}/K_I value is determined mostly by the inhibitor's **binding affinity** (K_I).

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E3 ligase assay: k_{inact}/K_I vs. IC₅₀ correlation

 \bullet The IC₅₀ values were determined in a separate end-point (not "continuous") assay.

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E3 ligase assay: k_{inact}/K_I vs. IC₅₀ order of potency

 \bullet The wrong "best" compound (B as opposed to A) was picked out by the IC $_{50}$ method.

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- \bullet For a set of "highly active" E3 ligase inhibitors, nearly all IC₅₀ values are essentially identical.
- \bullet However, the IC₅₀ method falsely singles out 'B' as the a "stand-out" compound:

$$
B \quad \gt{>} \quad A = C = D = E = F = G = H = I = J
$$

- \bullet In contrast, the k_{eff} values are much more distinct, which allows proper ranking.
- \bullet By the more accurate k_{eff} method, compound 'A' is most potent, closely followed 'B':

$$
A \geq B \quad > \quad C \geq D \geq E \quad > \quad F \geq G \quad > \quad H \geq I \quad > \quad J
$$

In the specific case of E3 ligase inhibition, $k_{\text{eff}} = k_{\text{inact}} / K_{\text{I}}$ is clearly superior to IC₅₀.

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Bioorg, Med. Chem. 29 (2021) 115865

The advantages of describing covalent inhibitor in vitro potencies by IC_{50} at a fixed time point. IC₅₀ determination of covalent inhibitors provides meaningful data to medicinal chemistry for SAR optimization

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- Two-step inhibitor binding ("rapid equilibrium" in the initial binding step).
- \bullet Identical to the mechanism also assumed for E3 ligase inhibition.

Thorarensen, A. et al. (2021) Bioorg. Med. Chem. 29, 115865

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Figure 1. Dose response illustration of the time dependency of IC_{50} for covalent inhibitors (model data).

• Article Abstract: "The potency of covalent inhibitors is generally considered to be more accurately described by the time-independent kinetic parameter k_{inact}/K_I rather than a by a simple IC_{50} , since the latter is a time-dependent parameter."

Thorarensen, A. et al. (2021) Bioorg. Med. Chem. 29, 115865

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JAK3 kinase: k_{inact}/K_I vs. IC₅₀ correlation (as published)

Figure 3. Relationship of k_{inact}/K_i with IC_{50} for JAK3 covalent inhibitors

Thorarensen, A. et al. (2021) Bioorg. Med. Chem. 29, 115865

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JAK3 kinase: k_{inact}/K_I vs. IC₅₀ correlation (digitized)

Digitization software: Engauge ver. 4

Thorarensen, A. et al. (2021) Bioorg. Med. Chem. 29, 115865

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JAK3 kinase: k_{inact}/K_I vs. IC₅₀ correlation (digitized and analyzed)

- About 20 percent of compounds are outside of half-order-of-magnitude band.
- The IC₅₀ for several compounds is "off" by more than two-orders of magnitude.

Thorarensen, A. et al. (2021) Bioorg. Med. Chem. 29, 115865

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JAK3 kinase: k_{inact}/K_I vs. IC₅₀ potency rank

NOTE: rank "1" is the "best" rank (highest potency) 140 130 o, a. 120 110 100 top 10% by 90 rank IC₅₀ 80 ú. Kinact / Ki 70 60 50 40 30 20 top 10% by Kinger/Ki 10 $\overline{0}$ 10 20 30 50 100 110 120 130 140 IC₅₀ rank

 \bullet The "top 10%" rule by IC₅₀ is only about 50% efficient: it misses 1/2 of true "10%" hits.

Even the "top 20%" rule by IC₅₀ misses compounds rank[ed](#page-25-0) [No. 1, 5, and 7 by](#page-0-0) k_{inact}/K_I k_{inact}/K_I k_{inact}/K_I k_{inact}/K_I k_{inact}/K_I k_{inact}/K_I **.**

Petr Kuzmič (BioKin Ltd.) **[Covalent IC50](#page-0-0)** 26 Sep 2024 27 / 1

Results

- The published log(IC₅₀) vs. log($k_{\rm inact}/K_{\rm I}$) correlation ($R^2\approx 0.75$) looks impressive.
- \bullet However, a comparison of the corresponding potency ranks tells a different story:
- **•** The IC₅₀ method is about 50% efficient in finding true hits ("top 10%" by k_{inact}/K_I).

Discussion

- An IC₅₀ (a single-point assay) is definitely cheaper than $k_{\rm inact}/K_{\rm I}$ (a continuous assay).
- The Big Question: Is this the usual "you get what you pay for" scenario?

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Pagliarini, R. A. et al. (2024) Nature Cancer, submitted.

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$STX-721$ mutant / w.t. selectivity by "apparent" efficiency constant

- "Apparent" efficiency constant: $k_{\text{eff}}^{(app)} = k_{\text{eff}} / (1 + [\text{ATP}]/K_{\text{m}}^{(ATP)})$
- \bullet Evaluated at $[ATP] = 1.0$ mM, similar to intracellular environment.

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STX-721 mutant / w.t. selectivity by "true" efficiency constant

 \bullet "True" efficiency constant $k_{\text{eff}} \equiv k_{\text{inact}} / K_{\text{I}}$ is uncorrected for cellular ATP.

 \bullet By this measure, STX-721 is still moderately selective (4x) for mutant over wild-type.

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STX-721 mutant / w.t. selectivity by "true" inhibition constant

 \bullet "True" inhibition constants K_I (binding affinities) are identical for mutant and wild-type!

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STX-721 mutant / w.t. selectivity by inactivation rate constant k_{inact}

• Selectivity for mutant largely derives from chemical reactivity of the warhead.

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STX-721 mutant / w.t.: inactivation rate constant k_{inact} ratios

The fictitious "compound numbers" largely reflect the progress of this project over time.

At some point, the team apparently stumbled upon a mu[tan](#page-33-0)[t-selective warhead](#page-0-0)[.](#page-0-0)

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 \bullet Mutant selectivity derives mostly from this warhead, or from its immediate surroundings.

 \bullet Binding affinity, due to the rest of the molecule, affects selectivity only to a minor degree.

Pagliarini, R. A. et al. (2024) Nature Cancer, submitted.

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- **The catalytic efficiency constant** k_{inact}/K_I **does not alwys tell the whole story.**
- **•** For two-step covalent inhibitors, we should always look at k_{inact} and K_{I} separately.
- In most projects, k_{inact} stays about the same while K_I gets lower during optimization.
- \bullet However, in the case of the EGFR 770-insNPG-771 mutant, k_{inact} determines selectivity.

- In designing corporate databases storing covalent ligand potency, make room for all of these:
	- k_{inact}/K_I : the covalent efficiency constant
	- \bullet k_{inact} : the inactivation rate constant
	- \bullet K_I: the inhibition constant (binding affinity)

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Discovery and Preclinical Characterization of BIIB129, a Covalent, Selective, and Brain-Penetrant BTK Inhibitor for the Treatment of **Multiple Sclerosis**

Published as part of Journal of Medicinal Chemistry virtual special issue "Exploring Covalent Modulators in Drug Discovery and Chemical Biology".

- Biogen recently released all relevant biochemical kinetic data to BioKin for publication. \bullet
- \bullet A manuscript is being prepared by BioKin $+$ Biogen, to describe all kinetics details.

Himmelbauer, M. K. et al. (2024) J. Med. Chem., 67, 8122-8140.

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