

Biochemical / Biophysical Kinetics "Made Easy"

Software DYNAFIT in drug discovery research

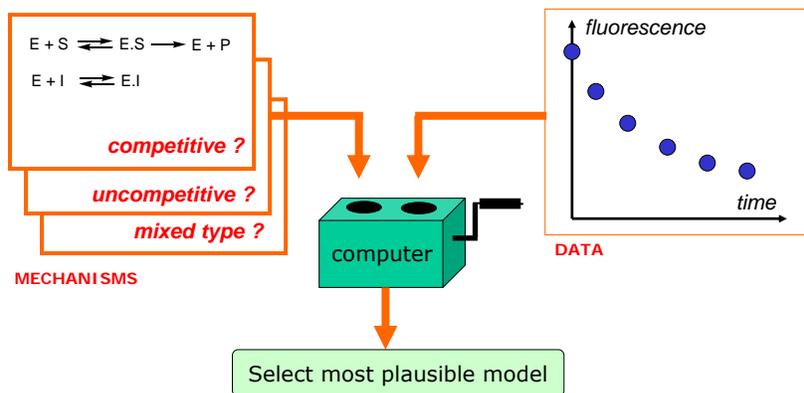
Petr Kuzmič, Ph.D.
BioKin, Ltd.

1. **Theory:** differential equation models
 - DYNAFIT software
2. **Example:** lanthaScreen® Eu assay in "kinetic" mode
 - p38a kinase / antibody / tracer
 - p38a kinase / antibody / tracer / **desatinib**



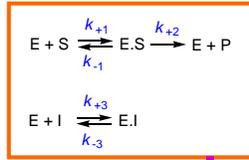
The task of mechanistic kinetics

SELECT AMONG MULTIPLE CANDIDATE MECHANISMS

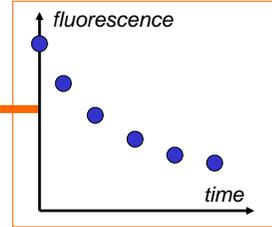


From mechanistic to mathematical models

DERIVE A MATHEMATICAL MODEL FROM BIOCHEMICAL/BIOPHYSICAL IDEAS



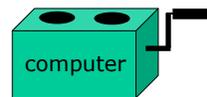
MECHANISM



DATA

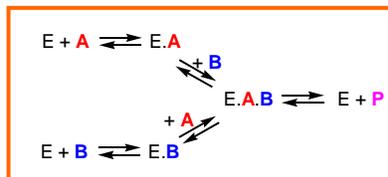
$$v = k_{+2}[E] \frac{k_{+1}k_{-3}[S]}{k_{-3}(k_{-1} + k_{+2}) + k_{-3}k_{+1}[S] + k_{+3}(k_{-1} + k_{+2})[I]}$$

MATHEMATICAL MODEL



Problem: Simple mechanisms ...

MERELY FIVE REACTIONS ...



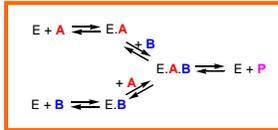
"RANDOM BI-UNI" MECHANISM

- 2 reactants (A, B)
- 1 product (P)
- 5 reversible reactions
- 10 rate constant

... lead to complex algebraic models

MERELY FIVE REACTIONS ...

Segel, I. (1975) *Enzyme Kinetics*. John Wiley, New York, p. 646.



"RANDOM BI-UNI" MECHANISM

$$\frac{v}{[B]} = \frac{K_1[A][B] + K_2[A]^2[B] + K_3[A][B]^2 - K_4[P] - K_5[A][P] - K_6[B][P]}{K_7 + K_8[A] + K_9[B] + K_{10}[A][B] + K_{11}[A]^2 + K_{12}[B]^2 + K_{13}[A][B]} + K_{14}[A][B]^2 + K_{15}[P] + K_{16}[A][P] + K_{17}[B][P] + K_{18}[A][B][P] \quad (IX-181)$$

$(k_{-5} = 0)$

where K_1 through K_{18} represent combinations of rate constants:

$$\begin{aligned}
 K_1 &= k_1 k_{-2} k_3 k_5 + k_{-1} k_2 k_4 k_5, & K_2 &= k_1 k_3 k_4 k_5, & K_3 &= k_2 k_3 k_4 k_5, \\
 K_4 &= k_{-1} k_{-2} k_{-3} k_{-5} + k_{-1} k_{-2} k_{-4} k_{-5}, & K_5 &= k_{-1} k_{-3} k_4 k_{-5}, \\
 K_6 &= k_{-2} k_3 k_{-4} k_{-5}, & K_7 &= k_{-1} k_{-2} k_{-3} + k_{-1} k_{-2} k_{-4} + k_{-1} k_{-2} k_5, \\
 K_8 &= k_1 k_{-2} k_{-3} + k_1 k_{-2} k_{-4} + k_1 k_{-2} k_5 + k_{-1} k_{-3} k_4 + k_{-1} k_4 k_5, \\
 K_9 &= k_{-1} k_2 k_{-3} + k_{-1} k_2 k_{-4} + k_{-1} k_2 k_5 + k_{-2} k_3 k_{-4} + k_{-2} k_3 k_5, \\
 K_{10} &= k_1 k_{-2} k_3 + k_{-1} k_2 k_4 + k_1 k_3 k_{-4} + k_2 k_{-3} k_4 + k_3 k_4 k_5, \\
 K_{11} &= k_1 k_{-3} k_4 + k_1 k_4 k_5, & K_{12} &= k_2 k_3 k_{-4} + k_2 k_3 k_5, & K_{13} &= k_1 k_3 k_4, \\
 K_{14} &= k_2 k_3 k_4, & K_{15} &= k_{-1} k_{-2} k_{-3} + k_{-1} k_{-4} k_{-5} + k_{-2} k_{-3} k_{-5}, \\
 K_{16} &= k_{-1} k_4 k_{-5} + k_{-3} k_4 k_{-5}, & K_{17} &= k_{-2} k_3 k_{-5} + k_3 k_{-4} k_{-5}, & K_{18} &= k_3 k_4 k_{-5}
 \end{aligned}$$



New approach: Numerical Kinetics

NO MORE ALGEBRA: LET THE COMPUTER DEAL WITH IT !



Theoretical foundations: *Mass Action Law*

RATE IS PROPORTIONAL TO CONCENTRATION(S)

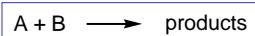
MONOMOLECULAR REACTIONS



rate is proportional to **[A]**

$$-d[A]/dt = k[A]$$

BIMOLECULAR REACTIONS



rate is proportional to **[A] × [B]**

$$-d[A]/dt = -d[B]/dt = k[A] \times [B]$$

Theoretical foundations: *Mass Conservation Law*

PRODUCTS ARE FORMED WITH THE SAME RATE AS REACTANTS DISAPPEAR

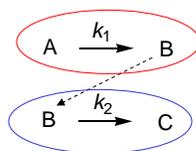
EXAMPLE



$$-d[A]/dt = +d[P]/dt = +d[Q]/dt$$

COMPOSITION RULE ADDITIVITY OF TERMS FROM SEPARATE REACTIONS

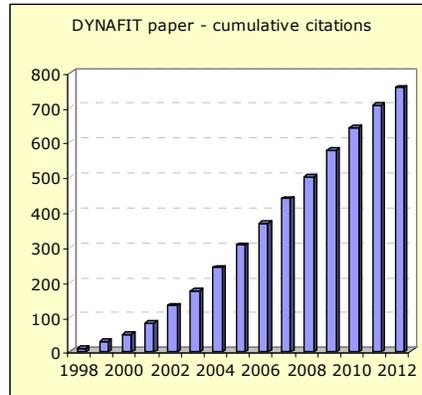
mechanism:



$$d[B]/dt = +k_1[A] - k_2[B]$$

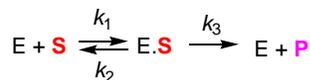
Software package DYNAFIT

- REFERENCES
1. Kuzmic P. (1996) *Anal. Biochem.* **237**, 260-273.
"Program DYNAFIT for the analysis of enzyme kinetic data"
 2. Kuzmic P. (2009) *Meth. Enzymol.*, **467**, 247-280.
"DYNAFIT – A software package for enzymology"

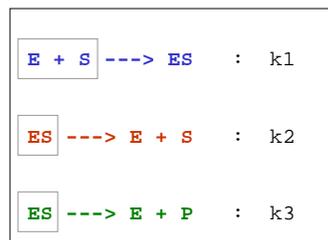


A "Kinetic Compiler"

HOW DYNAFIT PROCESSES YOUR BIOCHEMICAL EQUATIONS



Input (plain text file):



Rate terms:

$$k_1 \times [E] \times [S]$$

$$k_2 \times [ES]$$

$$k_3 \times [ES]$$

Rate equations:

$$d[E] / dt = -k_1 \times [E] \times [S]$$

$$+ k_2 \times [ES]$$

$$+ k_3 \times [ES]$$

$$d[ES] / dt = +k_1 \times [E] \times [S]$$

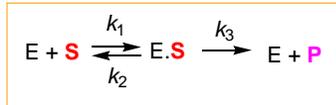
$$- k_2 \times [ES]$$

$$- k_3 \times [ES]$$

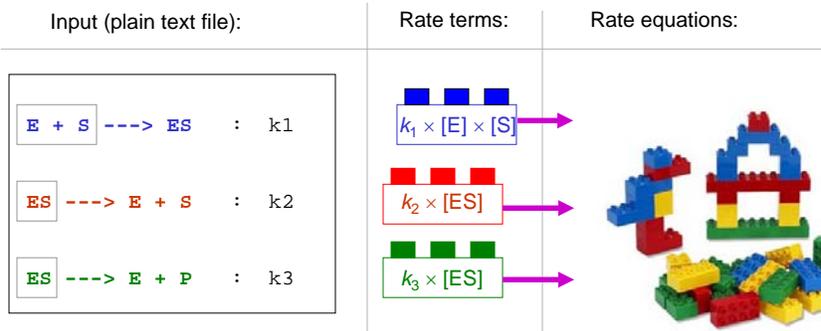
Similarly for other species...

System of Simple, Simultaneous Equations

HOW DYNAFIT PROCESSES YOUR BIOCHEMICAL EQUATIONS



"The **LEGO** method"
of deriving rate equations



DynaFit can analyze many types of experiments

MASS ACTION LAW AND MASS CONSERVATION LAW IS APPLIED IN THE SAME WAY

	EXPERIMENT	DYNAFIT DERIVES A SYSTEM OF ...
chemistry biophysics pharmacology — enzymology —	Kinetics (time-course)	Ordinary differential equations (ODE)
	Equilibrium binding	Nonlinear algebraic equations
	Initial reaction rates	Nonlinear algebraic equations

Biochemical / Biophysical Kinetics "Made Easy"

Software DYNAFIT in drug discovery research

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1. Theory: differential equation models

- DYNAFIT software

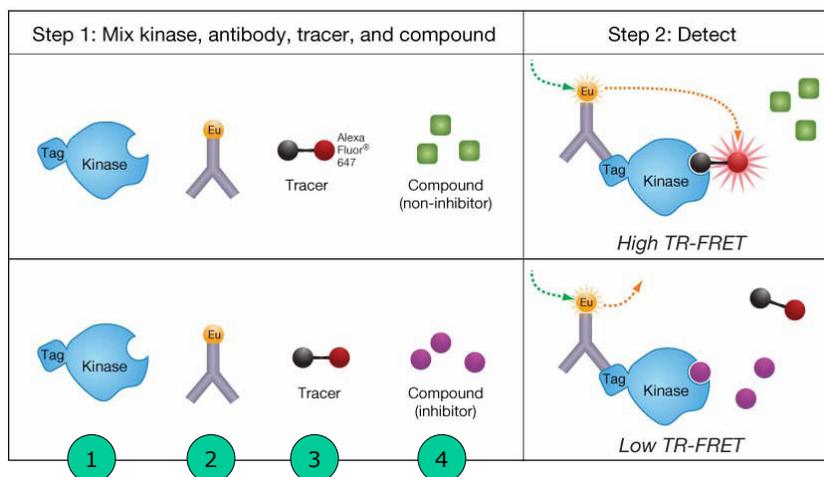
2. Example: lanthaScreen® Eu assay in "kinetic" mode

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- p38 α kinase / antibody / tracer / **desatinib**



Kinase – Antibody – Tracer – Inhibitor assay

A **FOUR-COMPONENT** MIXTURE

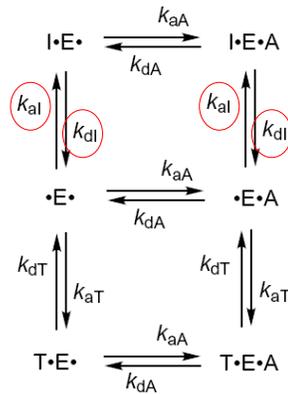


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Kinase – Antibody – Tracer – Inhibitor: mechanism

PURPOSE: OBTAIN RATE CONSTANTS FOR INHIBITOR ASSOCIATION & DISSOCIATION



E ... **e**nzyme
A ... **a**ntibody (FRET donor)
T ... **t**racer (FRET acceptor)
I ... **i**nhibitor

- **four** components
- **five** complexes (3 binary, 2 ternary)
- **six** *unique* rate constants

Rate constants and receptor-ligand residence time

IS IT WORTH CHASING AFTER RATE CONSTANTS?

Mbalaviele *et al.* (2009) *J. Pharm. Exp. Ther.* **329**, 14-25

"PHA-408 is an ATP competitive inhibitor, which binds **IKK-2** tightly with a relatively slow **off rate**."

Puttini *et al.* (2008) *haematologica* **93**, 653-61

"The present results suggest a slower **off-rate** (dissociation rate) of [a novel **Abi kinase** inhibitor] compared to **imatinib** as an explanation for the increased cellular activity of the former."

Tummino & Copeland (2008) *Biochemistry* **47**, 5481-92

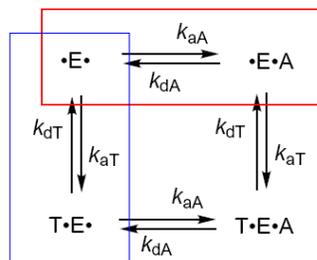
"... the extent and duration of responses to receptor-ligand interactions depend greatly on the **time period** over which the ligand is in **residence** on its receptor."

Let's look at **Kinase + Antibody + Tracer** only.

No Inhibitor yet.

Research plan: Stepwise model building

ASSUME THAT THE ANTIBODY AND THE PROBE BIND **INDEPENDENTLY**



1. get k_{aA} and k_{dA}

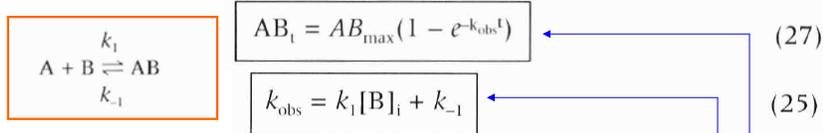
2. get k_{aT} and k_{dT}

• only then add the "unknown" inhibitor

Try to find conditions that might allow treating this as a simple "A + B" (two-component) system.

Classical method: “ $k^{(obs)}$ ” assuming [Antibody] \gg [Kinase]

GOODRICH & KUGEL (2006) “Binding and Kinetics for Molecular Biologists”, pages 91-95



Summary: How to experimentally measure k_1 and k_{-1}

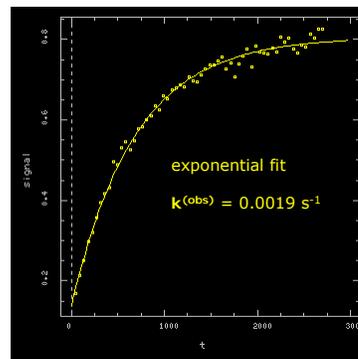
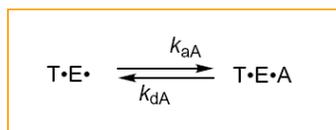
1. Develop an assay to monitor AB.
2. Combine A and B at $t = 0$ ($[B]_i \gg [AB]_{\max}$).
3. Measure accumulation of AB over time.
4. Plot data and determine k_{obs} (using Equation 24 or 27).
5. Repeat Steps 2–4 at multiple different $[B]_i$.
6. Plot k_{obs} versus $[B]_i$ and determine k_1 ^(and k_{-1}) (Equation 25).
7. Consider whether k_1 is diffusion limited.

Enzyme + Antibody at very large excess of [Tracer] (pt. 1)

THIS MIGHT ALLOW US TO TREAT THIS AS A SIMPLE “A + B” (TWO-COMPONENT) SYSTEM

METHOD:

1. Incubate Tracer & Kinase
 - [Tracer] fixed, very large excess
 - [Kinase] varied
2. Wait 10 minutes to equilibrate
3. Add [Antibody]
4. Measure increase in fluorescence (T.E.A)



[E] = 1.613 nM
[A] = 0.2 nM
[T] = 200 nM

Enzyme + Antibody at very large excess of [Tracer] (pt. 2)

DYNAFIT INPUT "SCRIPT": WE CAN USE SIMPLE ALGEBRAIC MODELS AS WELL

```
[task]

task = fit
data = generic

[parameters]

t
Ao, A, k

[model]

Ao = 0.1 ?
A = 1 ?
k = 0.001 ?

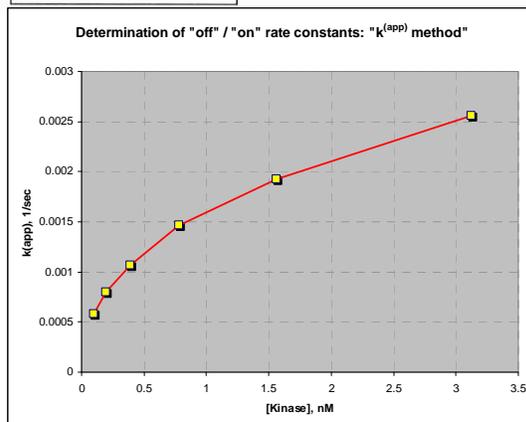
F = Ao + A*(1 - exp(-k*t))
```

$$AB_t = AB_{\max}(1 - e^{-k_{\text{obs}}t}) \quad (27)$$

Enzyme + Antibody at very large excess of [Tracer] (pt. 3)

TRY TO FIT k_{obs} TO THE STRAIGHT LINE MODEL EQUATION

$$k_{\text{obs}} = k_1[B]_i + k_{-1} \quad (25)$$



?!
not a straight line

Enzyme + Antibody at very large excess of [Tracer] (pt. 4)

POSSIBLE REASONS FOR THE **NONLINEARITY** OF k_{app} VS. [Kinase] PLOT

1. [Kinase] is *not* at very large excess over [Antibody] at all times

	A	B	C	D
assay #	[E], nM	[Ab], nM	[E] / [Ab]	
1	3.13	0.20	15.6	
2	1.56	0.20	7.8	
3	0.78	0.20	3.9	
4	0.39	0.20	2.0	
5	0.20	0.20	1.0	
6	0.10	0.20	0.5	

Summary: How to experimentally measure k_i

1. Develop an assay to monitor AB.
2. Combine A and B at $t = 0$ ($[B]_i \gg [AB]_{max}$).
3. Measure accumulation of AB over time.

Goodrich & Kugel (2006), p. 95

2. Signal in LanthaScreen® not strictly proportional to concentrations ?

The "emission ratio" is calculated as the 665 nm signal **divided** by the 615 nm signal.

3. Kinase concentrations being "off" their nominal values ?



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Enzyme + Antibody at very large excess of [Tracer] (pt. 5)

GLOBAL FIT TO A MECHANISTIC MODEL

```
[task]
  task = fit
  data = progress

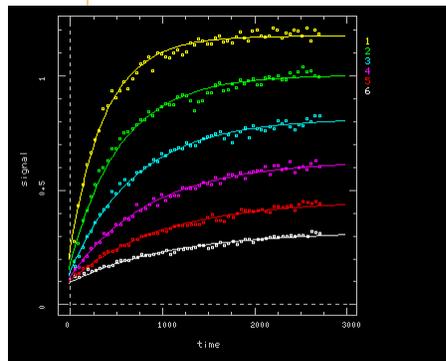
[mechanism]
  E + Ab <==> E.Ab : kaA kdA

[constants]
  kaA = 0.001 ?
  kdA = 0.001 ?

[concentrations]
  Ab = 0.2

[responses]
  E.Ab = 3 ?

[data]
  ...
  file d07 | concentration E = 3.1250 ?
  file d06 | concentration E = 1.5625 ?
  file d05 | concentration E = 0.7813 ?
  ...
```



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Enzyme + Antibody at very large excess of [Tracer] (pt. 6)

GLOBAL FIT TO A MECHANISTIC MODEL: BEST-FIT MODEL PARAMETERS

Optimized Parameters

No.	Par#Set	Initial	Final
#1	kaA	0.001	0.000920853
#2	kdA	0.001	0.000614225
#3	[E]#1	3.125	2.18796
#4	r(E.Ab)	3	7.23505
#5	offset	0.1	0.0858441
#6	[E]#2	1.5625	1.28789
#7	[E]#4	0.3906	0.478694
#8	[E]#5	0.1953	0.281575
#9	[E]#6	0.0977	0.162183

$$k_{\text{on}}^r = 0.92 \times 10^6 \text{ M}^{-1} \text{ s}^{-1}$$

$$k_{\text{off}}^r = 0.00061 \text{ s}^{-1}$$

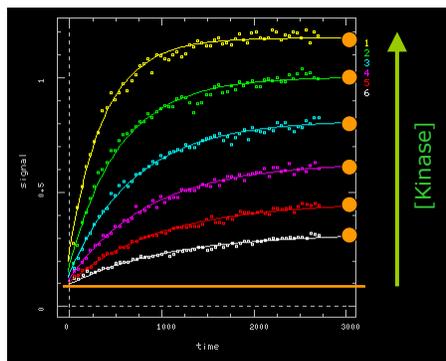
$$K_d = k_{\text{off}}^r / k_{\text{on}}^r = 0.66 \text{ nM}$$

[E], nM nominal	[E], nM best-fit	Ratio
3.13	2.19	1.4
1.56	1.29	1.2
0.78		
0.39	0.48	0.8
0.20	0.28	0.7
0.10	0.16	0.6

Signal in LanthaScreen® not strictly proportional to concentrations ?

Enzyme + Antibody at very large excess of [Tracer] (pt. 7)

FIT AN "EQUILIBRIUM" BINDING MODEL TO END-OF-TRACE SIGNAL VALUES



[task]

```
task = fit
data = equilibria
```

[mechanism]

$$E + Ab \rightleftharpoons E.Ab \quad : \quad KdA \text{ dissociation}$$

[constants] ; nM

$KdA = 0.7 ?$

...

$K_{dA} = 0.6 \text{ nM}$

Enzyme + Antibody at very large excess of [Tracer] (pt. 8)

SUMMARY

RESULTS

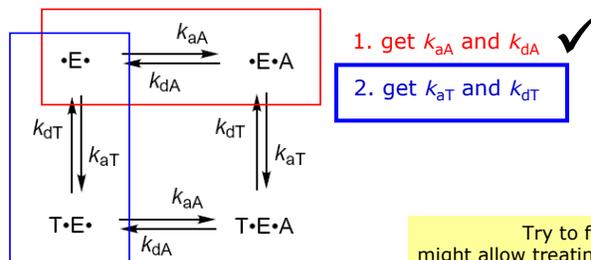
- Kinase–Antibody dissociation equilibrium constant is around **0.7 nM**
- The association rate constant is $0.9 \times 10^6 \text{ M}^{-1}\text{s}^{-1}$ (“diffusion control”)
- The half-time for dissociation is about **20 minutes** (“slow”)

CONCLUSIONS

- At **nanomolar** concentrations of [Kinase] and [Antibody] there is always a **mixture** of all three species (E, Ab, E.Ab)
- **Diluting** any kinase/antibody stock solution will cause the antibody to “**fall off**” the E.Ab complex
- This “**falling off**” of the antibody will be **slow, measured in minutes**

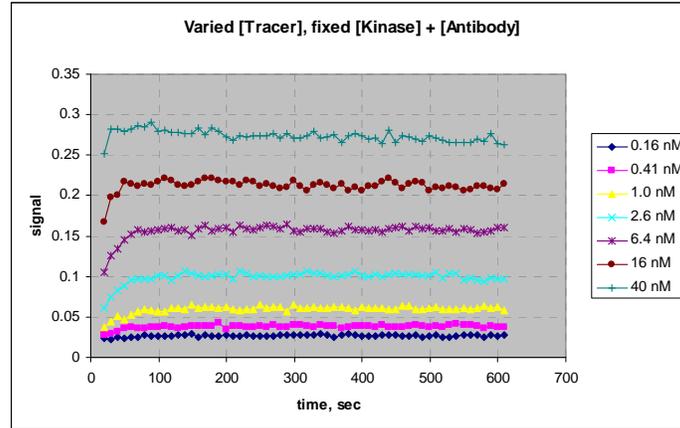
Research plan: Stepwise model building

ASSUME THAT THE ANTIBODY AND THE PROBE BIND INDEPENDENTLY



Kinase - Antibody - Tracer: varied [Tracer] (pt. 1)

RAW DATA



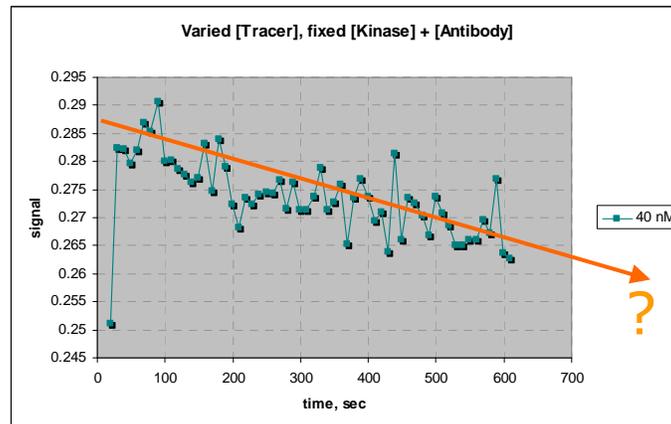
1. 5 μ L 1.5 nM Kinase + 6 nM Antibody
2. 5 μ L solvent
3. incubate 30 min
4. 5 μ L Tracer, varied final concentration



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Kinase - Antibody - Tracer: varied [Tracer] (pt. 1)

RAW DATA: CLOSER LOOK AT HIGH [TRACER] CONCENTRATION ASSAY



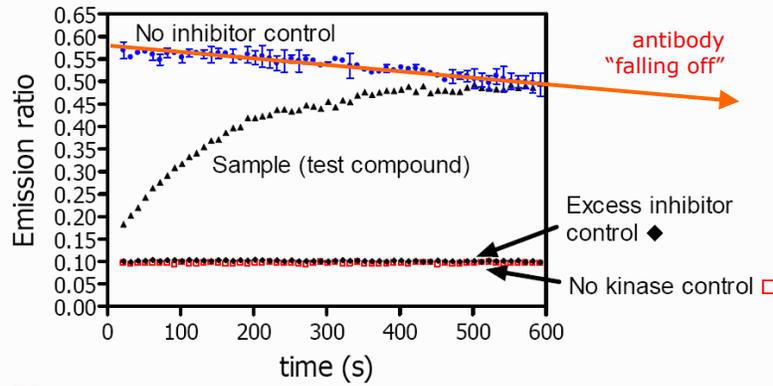
1. 5 μ L 1.5 nM Kinase + 6 nM Antibody
2. 5 μ L solvent
3. incubate 30 min
4. 5 μ L Tracer, 40 nM final concentration



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Gradual "loss of signal" upon dilution: Invitrogen literature

RESULTS FROM POSTER PRESENTED BY INVITROGEN



Determination of Drug-Kinase Residence Time in a Homogenous, Low-Volume Format

Bryan Marks, Connie Lobakken, Steve Riddle, Richard Somberg, Sara Horeley
OGE Systems Division • Life Technologies • 501 Chemistry Drive • Madison, Wisconsin 53719 • USA

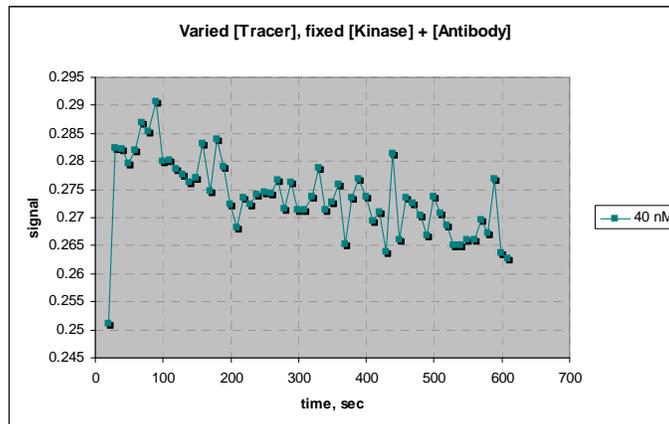


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Problem: Can't use "classical" method based on $k^{(app)}$

THIS IS DEFINITELY NOT A RISING EXPONENTIAL



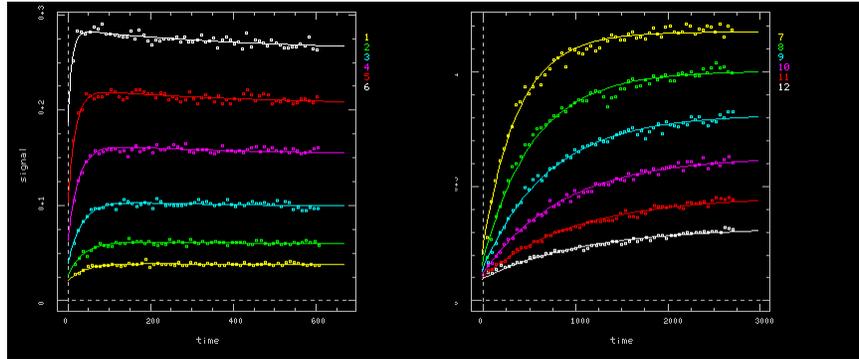
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Solution: Global fit of both sets of experiments (pt. 1)

EXPERIMENT #1: VARIED [TRACER]

EXPERIMENT #2: VARIED [KINASE]



[A] = 2 nM
[E] = 0.5 nM
[T] = 0.4 ... 40 nM

[A] = 0.2 nM
[E] = 0.098 ... 3.13 nM
[T] = 200 nM



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Solution: Global fit of both sets of experiments (pt. 2)

DYNAFIT INPUT ("SCRIPT") FILE

```
[task]
task = fit
data = progress
```

```
[mechanism]
```

```
E + Tr <====> E.Tr      : kaT kdT
E + Ab <====> E.Ab       : kaA kdA

E.Tr + Ab <====> E.Tr.Ab : kaA kdA
E.Ab + Tr <====> E.Tr.Ab : kaT kdT
```

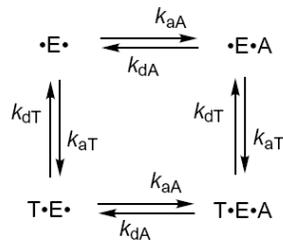
```
[constants] ; nM
```

```
kaA = 0.001 ?
kdA = 0.001 ?
```

```
kaT = 0.001 ?
kdT = 0.01 ?
```

```
...
...
...

```



Bio/Chemical Kinetics Made Easy

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Solution: Global fit of both sets of experiments (pt. 3)

DYNAFIT OUTPUT: BEST-FIT PARAMETERS

Optimized Parameters

No.	Par#Set	Initial	Final
#1	kaT	0.001	0.00265125
#2	kdT	0.01	0.0209196
#3	kaA	0.001	0.000920579
#4	kdA	0.001	0.00061413

$$k_{\text{on}}^{\text{T}} = 2.7 \times 10^6 \text{ M}^{-1} \text{ s}^{-1}$$
$$k_{\text{off}}^{\text{T}} = 0.021 \text{ s}^{-1}$$

$$K_{\text{dT}} = 7.9 \text{ nM}$$

$$k_{\text{on}}^{\text{A}} = 0.92 \times 10^6 \text{ M}^{-1} \text{ s}^{-1}$$
$$k_{\text{off}}^{\text{A}} = 0.00061 \text{ s}^{-1}$$

$$K_{\text{dA}} = 0.66 \text{ nM}$$

CONCLUSIONS

- Tracer is tight binding but "not that tight":
It will be **impossible to completely prevent inhibitor rebinding** by competition.
- Tracer is relatively "fast off" but "not that fast" ($t_{1/2} = 0.5 \text{ min}$):
In certain experiments there could be transients due to tracer interactions.
- It's a fairly complicated system:
Once inhibitor is added *all six rate constants* need to be considered.

Now for the complete four-component system:

Kinase + Antibody + Tracer + Inhibitor

Kinase - Antibody - Tracer - Inhibitor: data

KINASE: p38 α | **ANTIBODY:** anti-GST | **TRACER:** Invitrogen "Tracer-199" | **INHIBITOR:** desatinib

Data: Bryan Marks, Invitrogen

EXPERIMENT:

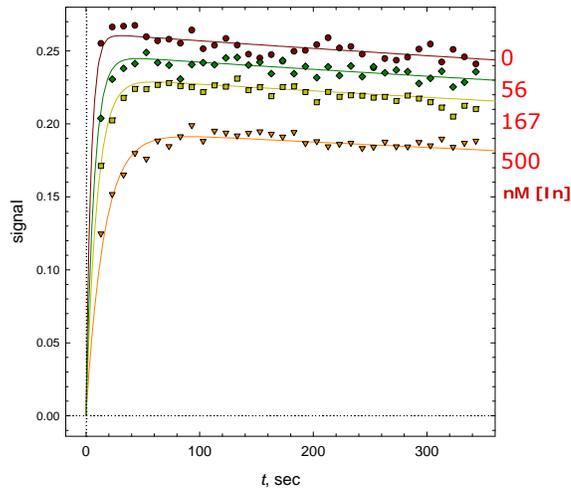
1. incubate

[E] = 4 nM
[Ab] = 40 nM
[In] = varied

30 minutes

2. dilute 1:20 with Tracer final concentrations

[E] = 0.2 nM
[Ab] = 2 nM
[Tr] = 100 nM
[In] = varied



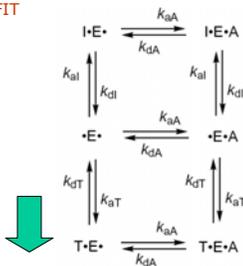
Bio/Chemical Kinetics Made Easy

37

Kinase - Antibody - Tracer - Inhibitor: fitting model

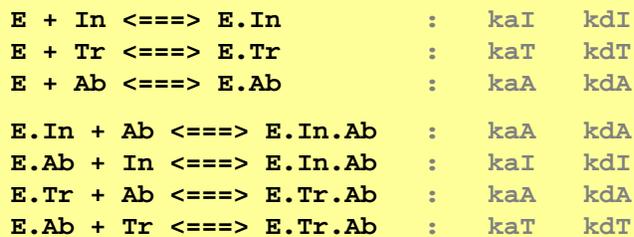
AUTOMATICALLY DERIVED BY DYNAFIT

system of simultaneous
ordinary
differential equations



[mechanism]

DynaFit Input



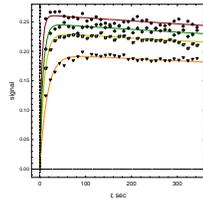
Bio/Chemical Kinetics Made Easy

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Kinase - Antibody - Tracer - **Inhibitor**: rate constants

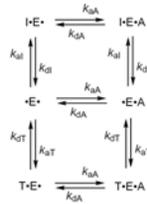
ASSUMPTION: **INDEPENDENT BINDING SITES** – ONLY **TWO ADDITIONAL RATE CONSTANTS**

DATA



+

MODEL



LEAST-SQUARES FIT

$$k_{aI} = 2.1 \times 10^9 \text{ M}^{-1} \cdot \text{s}^{-1}$$

$$k_{dI} = 19 \text{ s}^{-1}$$

PARAMETERS

"RESIDENCE TIME"

$$\tau = 0.05 \text{ sec}$$

Kinase - Antibody - Tracer - **Inhibitor**: state variables

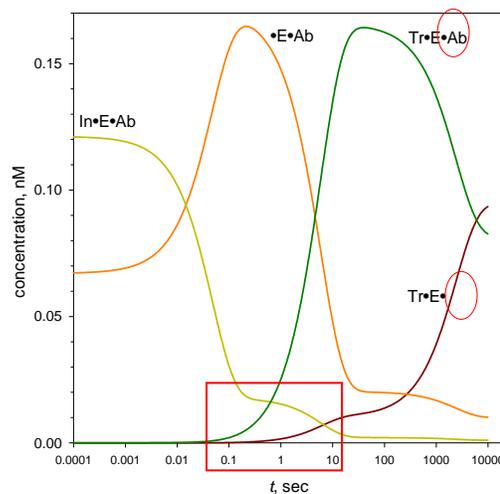
EVOLUTION OF SPECIES CONCENTRATIONS DURING THE KINETIC EXPERIMENT

EXPERIMENT:

- incubate
 $[E] = 4 \text{ nM}$
 $[Ab] = 40 \text{ nM}$
 $[In] = 370 \text{ nM}$
 30 minutes
- dilute 1:20 with Tracer
final concentrations
 $[E] = 0.2 \text{ nM}$
 $[Ab] = 2 \text{ nM}$
 $[Tr] = 100 \text{ nM}$
 $[In] = 18.5 \text{ nM}$



optimize design!



Acknowledgments

- **Bryan Marks:** all kinase experiments
Invitrogen, a.k.a. Life Technologies, Madison, Wisconsin



Questions ?

<http://www.biokin.com>